Preparation of silyl substituted crotylzinc reagents and their highly diastereoselective addition to carbonyl compounds[†]

Matthew D. Helm, Peter Mayer and Paul Knochel*

Received (in Cambridge, UK) 6th February 2008, Accepted 6th March 2008 First published as an Advance Article on the web 19th March 2008 DOI: 10.1039/b802157k

Readily prepared β -silyl substituted crotylzinc reagents undergo highly selective allylation of carbonyl compounds leading to *syn*-homoallylic alcohols.

Allylation reactions of carbonyl compounds with allylic organometallic reagents are of huge importance in organic synthesis.¹ The allylation reaction is one of the best methods for the controlled synthesis of quaternary centres. Recently we have reported the use of zinc dust in the presence of lithium chloride in THF as a cheap and convenient method for the synthesis of substituted allylic zinc reagents from allyl chlorides or phosphonates.² Allylation reactions with carbonyl compounds at -78 °C provided mostly the *anti* isomers. To extend this methodology to the preparation of syn-homoallylic alcohols we have investigated the possibility of using steric hindrance on the 2-position of the allyl chloride, through the presence of a silyl group. The resulting allylic zinc reagents should undergo highly syn-selective allylation of aldehydes and ketones (Fig. 1). The use of silvl substituted allyl-chromium reagents has been investigated previously, however, moderate levels of diastereoselectivity were observed.³

We decided to concentrate our initial studies using a mixture of readily available trimethylsilyl substituted allyl chlorides⁴ 1a and 1b (Scheme 1). In the presence of zinc powder (10 equiv.) and LiCl (3 equiv.), we observed a smooth insertion reaction in which complete consumption of the starting materials was observed in 18 h at 25 °C and provided organozinc species 2 in 78% yield as determined by iodometic titration. Remarkably, addition of 1.1 equiv. of this reagent to benzaldehyde (1 equiv.) at -78 °C gave the allylation product 3 as a single diastereomer as distinguishable by ¹H-NMR. Protodesilvlation of the allylation product 3 via a carbon to oxygen silyl migration promoted by sodium hydride in THF/HMPA⁵ gave the expected syn-homoallylic alcohol 4 in 86% yield (Scheme 1). The remarkable diastereoselectivity observed in this reaction must be attributed to the fact that only a single allylic zinc species (E-isomer) is formed during the insertion. This is highly unusual as previous examples of conformationally stable allylzinc reagents have relied upon either conformational rigidity, coordination or stereoelectronic factors.^{1,2,6}



A variety of alkyl substituted aldehydes were investigated (5–7, Fig. 2). In all cases, we were delighted to observe very high diastereoselectivity and the steric properties of the aldehyde appeared to have no bearing on the diastereoselectivity. Reaction with functionalised benzaldehyde derivatives showed a high level of tolerance towards sensitive functional groups such as free amino, chloro and nitro groups (8–9, Fig. 2). In the cases involving nitro substituted, more electron deficient, aldehydes lower diastereoselectivity was observed (9, Fig. 2 and 15–16, Scheme 2). The addition to acetophenones was also highly selective and again the tolerance towards sensitive functional groups was remarkable, with substrates containing azide and bromo functional groups proceeding smoothly (10 to 12, Fig. 2).

Reaction of 1.1 equiv of **2** with *p*-nitrobenzaldehyde gave homoallylic alcohol **15** in 89% yield with a diastereoselectivity



Ludwig-Maximilians-Universität München, Department Chemie und Biochemie, Butenandtstr. 5-13, Haus F, D-81377 München, Germany. E-mail: Paul.Knochel@cup.uni-muenchen.de; Fax: (+)49-89-2180-77680; Tel: (+)49-89-2180-77681

[†] Electronic supplementary information (ESI) available: Further experimental details; crystallographic data in CIF format (CCDC 673535). See DOI: 10.1039/b802157k



Fig. 2 Addition of trimethylsilyl substituted allyl zinc 2 to functionalised aldehydes and ketones.⁷



of 89 : 11 (Scheme 2). Careful recrystallisation from ethyl acetate and pentane gave only the major isomer which was identified as the *cis* isomer through X-ray analysis.‡ Formation of the triethylsilyl substituted allylic zinc 14 also proceeded smoothly (Scheme 2). Addition of 14 (1.1 equiv.) to *p*-nitrobenzaldehyde (1 equiv.) gave the allylation product 16 (Scheme 2) in 80% yield with a diastereoselectivity of 94 : 6. This increase in stereoselectivity may be best explained by the increased steric hindrance of the silyl group of 14 compared to 2 (Scheme 2).

In conclusion, we have demonstrated that the new β -silyl substituted allylic zinc reagent **2** undergoes highly diastereoselective additions to acetophenones and aldehydes. Further extensions of this methodology are currently under way in our laboratory.

We would like to thank the SFB (749) and the Alexander von Humboldt foundation for financial assistance and Umicore AG (Angleur, Belgium) for the generous gift of zinc dust.

Notes and references

‡ CCDC 673535.

- For allyl metal additions see: S. R. Chemler and W. R. Roush, in Modern Carbonyl Chemistry, ed. J. Otera, Wiley-VCH, Weinheim, 2000; S. E. Denmark and N. G. Almstead, in Modern Carbonyl Chemistry, ed. J. Otera, Wiley-VCH, Weinheim, 2000; Stereoselective Synthesis, Methods of Organic Chemistry (Houben-Weyl), ed. G. Helmchen, R. W. Hoffmann, J. Mulzer and E. Schaumann, Thieme, Stuttgart, E21 edn, 1996; I. Marek and G. Sklute, Chem. Commun., 2007, 1683; M. Yasuda, K. Hirata, M. Nishino, A. Yamamoto and A. Baba, J. Am. Chem. Soc., 2002, 124, 13442; A. N. Thadani and R. A. Batey, Org. Lett., 2002, 4, 3827; S. W. Li and R. A. Batey, Chem. Commun., 2004, 1382; C. T. Buse and C. H. Heathcock, Tetrahedron Lett., 1978, 1865; Y. Yamamoto, H. Yatagai, Y. Naruta and K. Maruyama, J. Am. Chem. Soc., 1980, 102, 7107.
- 2 H. Ren, G. Dunet, P. Mayer and P. Knochel, J. Am. Chem. Soc., 2007, 129, 5376.
- 3 D. M. Hodgson and C. Wells, *Tetrahedron Lett.*, 1992, **33**, 4761; S. Nowotny, C. E. Tucker, C. Jubert and P. Knochel, *J. Org. Chem.*, 1995, **60**, 2762; see also the use of silyl substituted allylstannanes: Y. Nishigaichi, N. Ishida, M. Nishida and A. Takuwa, *Tetrahedron Lett.*, 1996, **37**, 3701.
- 4 T. H. Chan and B. S. Ong, J. Org. Chem., 1978, 43, 2994; T. H. Chan, E. Mychajlowshij, B. S. Ong and D. N. Harpp, J. Org. Chem., 1978, 43, 1526.
- 5 F. Sato, M. Kusakabe and Y. Kobayashi, J. Chem. Soc., Chem. Commun., 1984, 1130.
- 6 F. Lambert, B. Kirschleger and J. Villiéras, J. Organomet. Chem., 1991, 405, 273; F. Lambert, B. Kirschleger and J. Villiéras, J. Organomet. Chem., 1991, 406, 71.
- 7 To further examine the scope of this methodology some further experiments were conducted. Reaction of **2** with 4-formyl-benzoic acid methyl ester gave the allylation product in 88% yield with a diastereoselectivity of 92 : 8. Reaction of **2** with *p*-anisaldehyde gave an unstable product which decomposed upon attempted isolation. Reaction of **2** with 4-acetyl-benzaldehyde gave a complex mixture of products containing the aldehyde allylation products and the α -addition product as the major contingent with traces of ketone allylation products. See ESI† for further details.